DETERMINATION OF THE GEOMETRICAL CONFIGURATION OF THE POLYEN CHAIN OF MONOCIS C40-CAROTENOIDS—I

A ¹³C NMR STUDY OF MONOCIS ZEAXANTHINS AND MONOCIS CAPSORUBINS

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Abstract—¹³C NMR studies reveal that the polyen chain of *neo*-zeaxanthin A (2) and that of *neo*-capsorubin A (5) have 13-cis configuration, while those of *neo*-zeaxanthin B (3) and *neo*-capsorubin B (6) have 9-cis geometrical configuration. ¹³C NMR data on the corresponding *all-trans* carotenoids (1 and 4) are also reported.

In the field of structural research into the geometrical isomers of isoprenoid compounds a new powerful method, ¹³C NMR spectroscopy, has recently been introduced and successfully used in the structural study of the *cis* isomers of retinal^{1,2} and more recently in the configurational assignment of violeoxanthin.³

The present "C NMR study of cis-carotenoids was begun with the investigation of zeaxanthin and capsorubin (symmetrical types), whose monocis isomers are known in the literature.46 Tentative spacial structures were suggested for their neo-A and neo-B forms by Zechmeister et al.;⁴ neo-zeaxanthin A, 2 (m.p. 98°; λ_{max} (ϵ_{max}) in benzene: 487 (70.200), 458 (86.300) and 347 (39.000) nm)[†] and neo-capsorubin A, 5 (m.p. 84°; λ_{max} (ϵ_{max}) in benzene: 517 (71.000), 483 (86.000) and 378 (40.000) nm)† with high cis-peaks represent 13-cis isomers, while *neo*-zeaxanthin B, 3 (m.p. 107°; λ_{max} (ϵ_{max}) in benzene: 488 (94-700), 458 (111-700) and 348 (8-800) nm)* and neocapsorubin B, 6 (m.p. 120°; λ_{max} (ϵ_{max}) in benzene: 519 (99.000), 485 (111.000) and 380 (16.000) nm)† with low cis-peaks represent 9-cis isomers. We found that "C NMR data gave exact evidence for the location of the cis double bonds in these molecules, confirming the earlier hypothetical assignments.

Natural abundance "C NMR spectra were obtained at 25-16 MHz using a Varian XL-100-15 spectrometer equipped with Varian S124-XL Fourier transform accessory and a 620L computer. The samples were prepared as 0-04 to 0-05 M solutions in CDCl₃ with TMS as internal reference. The probe temperature was maintained at 32°. Because of its poor solubility, *all-trans* zeaxanthin (1) was previously transformed into 3,3'-diacetate derivative. Depending on the sample from 6000 to 16,000 transients were accumulated with acquisition time 0-8 sec, spectral digitization 1-25 Hz/point and pulse flip angle 25-30°. The chemical shift values are believed to be accurate to ± 0.04 ppm.

The "C chemical shift values for the compounds studied are given in Table 1. The individual resonance values for the all-trans zeaxanthin (1) were assigned on the analogy of the data published for β -carotene' and taking into account changes due to the presence of the 3-OAc function. In addition to information derived from standard FT procedures, the designation of the carbon resonances in the all-trans capsorubin (4) was further facilitated by the alternating shifts of sp² resonances due to the polarizing effect of the carbonyl end group: With respect to their values in all-trans zeaxanthin (or β -carotene), even numbered polyenic carbons exhibited upfield shifts, while odd numbered carbons downfield shifts in a similar way as noted in a recent study on apocarotenoids.7 The magnitude of these shifts decreased with the number of bonds separating the relevant carbons from the carbonyl group.*

The cis-configured double bonds in the isomers (2, 3, 5 and 6) were located on the basis of known cis-trans isomerization shifts.^{1,2} The appearance of a Me resonance, 19 or 20Me, which was shifted downfield by about 8 ppm with respect to its value in the corresponding all-trans isomer (12.8 ppm), suggested 9- or 13-cis double bonding in each case. (Isomerization at position 11 is known to be evidenced by a lesser downfield shift of the 20Me signal^{1,2} while the occurrence of a *cis* double bond at 15 should give rise to a symmetrical spectrum differing from that of the parent all-trans isomer.) Minor steric shifts detected for some of the sp³ resonances and sizable resonance doublings of the signals due to carbons 5 and 6 in 3 or to the carbonyl group in 6 indicated that in these molecules the isomerization occurred at position 9. A more detailed evaluation of the isomerization shifts in the molecules studied provided corroboration for this qualitative assignment. The chemical shifts or the cis end carbon resonances relative to the corresponding trans end values are summarized in Table 2. Also presented in Table 2 are the chemical shift changes relative to the parent all-trans molecules (in parentheses); for the zeaxanthin isomers (2 and 3) many of the latter values are affected by the

[†]The data refer to the isomers obtained by iodine catalysis⁴ in our laboratory.





Table 1. ¹³C chemical shifts*

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Compound							1				Position
			l	2	3	4	5	6	7	8	9
<u>All-trans</u> Zeaxanthin- -3,3 ¹ -diacetate			36.73	44.24	68.47	38.57	125.66	138.03	125.36	138.73	135.55
Nec-Zeaxanthin B	trans	end	37.16	48.64	65.16	42.71	126,22	137.90	125.68	138.56	135.67
/9- <u>ois</u> -Zeaxanthin/	<u>ais</u>	end	37.16	48.64	65.16	42.71	126.47	138.15	127.52	130.80	134.28
Neo-Zeaxanthin A	trans	end	37.18	48 .6 6	65.15	42.72	126.37	137.92	125.71	138.58	135.63
/13- <u>cis</u> -Zeaxanthin/	<u>c1s</u>	end	37.18	48.66	65.15	42.72	126.26	137.92	125.71	138.58	135.71
<u>All-trans</u> Capsorubin			44.02	51.03	70.35	45.46	59.01	203.00	121.27	146.87	134.10
Neo-Capsoruhin B	trans	end	44.03	50.95	70.40	45.41	59.00	202.98	121.17	146.87	133.98
/9- <u>cis</u> -Capsorubin/	<u>018</u>	end	44.03	50.95	70.40	45.41	9 9.09	203.55	122.58	138.10	132.15
Neo-Capsorubin 🛦	<u>trans</u>	end	44.00	50.93	70.37	45.36	59.01	203.00	121.19	146.85	133.97
/13- <u>cis</u> -Capsorubin/	<u>çis</u>	end	44.00	50.93	70+37	45.36	59.01	203.00	121.55	146.75	134.07

a In ppm as measured from internal TMS in deuterochloroform solution

^b For these resonances two, clearly separated lines were observed, with shift differences of approx. 0.1 ppm. remains to be clarified. The figures reported in this Table represent mean values.



of isomeric carotenoids

											Other Resonances	
10	11	12	13	14	15	16Me	1716	1.8Me	19 Me	2014e	000 <u>Me</u>	0 <u>00</u>
131,52	124.96	137.74	136.46	132.68	130.18	28,58	30.07	21.40	12.75	12.75	21.39	170.63
131.38	124.98	1 <i>3</i> 7.65	136.38	132.63	130.05	28.84	30.34	21.59	12.83	12.83		
129.91	123.79	136.38	136.90	132.49	1 30.1 4	28.84	30.54	21.69	20.70	12.83		
131.38	124.96 ^b	137.70	156.43 ^b	132.57 ^b	130.19	28.82	30.3 4	21.59	12.78	12.78		
131.38	126.05	129 . 43 ^b	134.92	131-02	128.93	28.82	30.54	21.59	12.78	20.68		
140.50	124.67	141.75	137.00	154.94	131.25	25.14	25.94	21.37	12.83	12.83		
140.59	124.54	141.87	136.70	135.07	130.86	25.13	25.90	22.95	12.84	12.84		
139.09	123.70	140.59	137.26	134.95	151.42	25.10	25.90	21.27	20.44	13.00		
140.38	124.52	141.84	136.73	194.82	151.05	25.12	25.93	21.34	12.87	12.87		
140.58	125.88	133.38	135.24	135.28	129,90	25.12	25.93	21.34	12.87	20.61		and the second

Since after recording of the spectra no decomposition product could be identified, the origin of these splittings

Carbon		Сарвог	rubin		Zeaxantoin			
	9- <u>ci</u> s	<u>a</u> /Neo B/	13- <u>-c18</u>	/Neo A/	9- <u>c1s</u> /Neo B/	13- <u>cis</u> /Neo A/		
10	0.0	/0.0/	0.0	/0.0/	0.0	υ.0		
20	0.0	/0.1/	0.0	/0.0/	0.0	0.0		
<u>3</u> C	0.0	/0.0/	0.0	/0.0/	0.0	0.0		
4C	0.0	/0.0/	0.0	/0.0/	0.0	0.0		
5C	0.1	/0.1/	0.0	/0.0/	0.2	-0.1		
6C	0.6	/0.6/	0.0	/0.0/	0.2	0.0		
70	1.4	/1.3/	0.4	/0.3/	1.8	0.4		
8C	-8.8	/-8.8/	-0.1	/-0.1/	-7.8	0.2		
9C	-1.8	/-1.9/	0.1	/0.0/	-1.4	0.2		
100	-1.5	/-1.4/	0.2	/0.1/	-1.5	-0.1		
110	-0.8	/-1.0/	1.4	/1.2/	-1.2	1.1		
120	-1.3	/-1.2/	-8.5	/-8.4/	-1,3	-8.2		
130	0.6	/0.2/	-1.5	/-1.8/	0.5	-1.5		
14C	-0.7	/-0.6/	-1.5	/-1.7/	-0.2	-1.6		
150	0.6	/0.2/	-ì.2	/-1.4/	0.1	-1.3		
1610	0.0	/0.0/	0.0	/0.0/	0.0	0.0		
17 1 4e	0.0	/0.0/	0.0	/0.0/	0.0	0.0		
1810	-0.1	/-0.1/	0.0	/0.0/	0.1	0.0		
1914e	8.4	/8.4/	0.0	/0.0/	7.9	0.0		
2014e	0.2	/0.2/	7.8	/7.8/	0.0	7•9		

Table 2. Shifts of *cis*-end carbons relative to *trans*-end carbons, $\Delta \delta^{*,h}$

 $\frac{a}{\Delta \delta} = \delta_{\underline{cis}} - \delta_{\underline{trans} / in ppm/}$

^b Shifts relative to <u>all-trans</u> values in parentheses

acetylation shifts and therefore omitted. The relative shifts observed in the present study are in good agreement with the values reported for model compounds.¹²

It may be noted that the consistency in delineation of the *cis* and *trans* end resonances in the spectra of the *cis* isomers is closely intercorrelated with the correctness of assignment of the spectrum of the parent *all-trans* molecule. Such a consistency of our results suggests that the assignment of carbon 9 and that of carbon 13 of β -carotene are interchanged in Ref. 1 (see also Ref. 8).

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- ⁶As compared with the values in β -carotene, the following relative shifts were observed: 9C: 9·8; 10C: 9·6; 12C: 4·4; 14C: 2·5; 7C: -4·4; 9C: -2·4; 11C: -0·4, and 13C: -0·3 ppm. For the sake of consistency, shift values for 9C and 13C of Ref. 1 are interchanged (see text). Both spectra were recorded in CDCl₃ solutions.